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Study Skills in the Skills and Reference Handbook the end of your textbook. The study of living things Science From Curiosity (pages 2-3) 1. Define science. 2. The questions that to scientific are provided by 3.

Section 1.1 What Is Science?

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shape Variable volume Definite volume
Definite shape Solids Liquids Gases false
true plasma gas Particles in a solid are
packed close together in an orderly
arrangement. The arrangement of
particles in a liquid is more random. A
gas takes the shape and volume of its
container.

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Chapter 3 States of Matter Section 3.1 Solids, Liquids ...

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Physical Science Guided Reading and Study Workbook Chapter 12. Name _____ Class _____
Date _____ Chapter 12 Forces and Motion
Section 12.2 Newton's First and Second Laws of Motion (pages 363-369) This section discusses how force and mass affect acceleration. The acceleration due to gravity is defined, and mass and weight are compared.

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Physical Science Guided Reading and
Study Workbook Chapter 3 23 Solid
Liquid Liquid Solid Liquid Gas Gas Liquid
Solid Gas Gas Solid A phase change is
the reversible physical change that
takes place when a substance changes
from one state of matter to another. d a

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e f c b The temperature of a substance remains constant during a phase change. false

Chapter 3 States of Matter Section 3.3 Phase Changes

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Workbook Answer Key

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that can differ from those of its solute and solvent are conductivity, freezing point, and boiling point. During the formation of a solution, energy is either released or absorbed. • Before a solution can form, the attractions among the particles of solvent and of solute must be broken.

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Answers Chapter 1 **Chapter 8 Solutions, Acids, and Bases**

Physical Science Reading and Study
Workbook Chapter 2 19 Copper roof
changing color from red to green when
exposed to moist air Formation of
carbon dioxide gas when vinegar is
added to baking soda Formation of
cottage cheese curds when acid is

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added to milk. false reactivity A
chemical property is any property that
produces a change in

Chapter 2 Properties of Matter **Section 2.3 Chemical Properties**

Reading Skill Practice When reading
about related processes, drawings can
help you appreciate their similarities and

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differences. Study Figures 6, 7, and 8 in Section 11-1 and explain how the figures are similar and how they are different. Do your work on a separate sheet of paper.

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Give an example of Newton's first law of motion. Determine the speed if an object moves 25 meters in 19 seconds left. Determine the distance if the speed is 15 m/s and the object moved for 25 seconds. Determine the acceleration if a car has a velocity at the start of 10 m/s and it accelerates to 60 m/s in 5 seconds.

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Chapter 4 Atomic Structure Summary
4.1 Studying Atoms •The ancient Greek
philosopher Democritus believed that all
matter consisted of extremely small

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particles that could not be divided.

- Aristotle did not think there was a limit to the number of times matter could be

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